## Synthesis, and Cytotoxic Activity of $N^{\text{ind}}$ -Alkoxy Derivatives of Antibiotic Arcyriarubin and Dechloro-rebeccamycin Aglycon

S. A. Lakatosh, J. Balzarini, G. Andrei, R. Snoeck, E. De Clerco and M. N. Preobrazhenskaya\*

G.F. Gause Institute of New Antibiotics, Russian Academy of Medical Sciences, B. Pirogovskaya 11, Moscow 119021, Russia †Rega Institute for Medical Research, Katholieke Universiteit Leuven, B3000, Leuven, Belgium

(Received for publication April 4, 2002)

Interest in rebeccamycin and arcyriarubin derivatives comes from their interesting biological properties.

Rebeccamycin<sup>1)</sup> (1, Fig. 1) and some of its analogs have shown remarkable activity as DNA topoisomerase I inhibitors and antitumor compounds.2) Some of N-glycosyl derivatives of arcyriarubin A (e.g. 2) also have demonstrated potent antiproliferative activities<sup>3)</sup> N<sup>ind</sup>-alkyl derivatives of arcyriaflavin A (3) were shown to effect a potent and selective inhibition of human cytomegalovirus (HCMV) replication.<sup>4)</sup> Although various derivatives of indolo[2,3-a]carbazole and bisindolylmaleimide have been synthesized,2) the derivatives bearing N-alkoxy substituents have not been yet described. Our objective was to develop methods of synthesis of unsymmetrical *N*-alkoxy derivatives of dechlororebeccamycin aglycon and arcyriarubin and to study the impact of this type of substituent on cytotoxic and antiviral activities. The starting bisindolylmaleimide 4 (N-methyl arcyriarubin) was prepared from indole Grignard reagent *N*-methyl-2,3-dibromomaleimide previously as described.5) From the reaction mixture after column chromathography additionally to compound 4 (67%),

Fig. 1.

Scheme 1. Synthesis of *N*-methylarcyriarubin A (4) and derivative 5.

<sup>\*</sup> Corresponding author: lcta@space.ru.

compound 5 was isolated in 7.4% yield (Scheme 1). It could be formed from the indolenine intermediate<sup>5)</sup> and an excess of indolylmagnesium bromide, with subsequent oxidative aromatization (5, Rf 0.19 [(CHCl<sub>3</sub> - EtOAc, 6:1), a dark red solid, H<sup>1</sup>-NMR (DMSO- $d_6$ ,  $\delta$  ppm, J Hz) 2.98 (3H, s), 6.58 (1H, t, J=6.5), 6.82 (2H, t, J=7.41), 6.94 (1H, t, J=6.5)t, J=7.17), 7.02 (2H, t, J=7.39), 7.1 (1H, d, J=7.63), 7.14 (1H, d, J=7.14), 7.28 (1H, d, J=4.8), 7.29 (1H, d, J=2.9),7.41 (1H, d, J=7.24), 7.43 (1H, d, J=7.9), 7.65 (1H, d, J=8.05), 7.68 (1H, d, J=2.9), 11.29 (1H, d, J=2.2), 11.53 (1H, s), 11.59 (1H, d, J=2.62); <sup>13</sup>C-NMR (DMSO- $d_6$ ) 23.9 (N-CH<sub>3</sub>), aromatic CH 111.1, 111.6, 111.7, 119.1, 119.2, 119.3, 119.5, 120.6, 120.9, 121.4, 121.5, 125.0, 129.5, aromatic C 102.0, 106.1, 108.1, 125.2, 125.3, 126.8, 128.1, 131.5, 132.2, 134.9, 136.0, 136.1, carbonyl 171.1, 171.7. HR-MS calcd for  $C_{29}H_{20}N_4O_2$  M<sup>+</sup> 456.1586, found 456.1557.]

The key precursor in the synthesis of the *N*-methoxy-rebeccamycin aglycon analog, indolylindoline **7** was prepared from bisindolylmaleimide **4** *via* bisindolyl-succinimide **6** by treatment with TFA.<sup>6)</sup> Compound **7** was converted into the corresponding *N*-hydroxy derivative **8** by the oxidation with  $H_2O_2$  in the presence of catalytic amounts of  $Na_2WO_4\times 2H_2O$  as it was first described for the synthesis of 1-hydroxyindole.<sup>7)</sup> Subsequent methylation led

Table 1. Cytotoxicity values for compounds 5, 9, 10, and 13~20.

	Cytotoxicity, IC <sub>50</sub> <sup>a</sup> , μM					
#	L1210	MOLT4/C 8	CEM			
5	2.3±0.9	2.7±1.2	3.4±2.5			
9	100±30	13±3	65±40			
10	≥500	>500	>500			
13a	3.5±0.1	1.7±0.04	1.9±0.01			
13b	11 ± 2	$3.2 \pm 0.03$	$3.7 \pm 0.08$			
13c	0.51 <u>+</u> 0.01	0.25	$0.38 \pm 0.03$			
14a	32 ± 2	26 ± 2	17 ± 1			
14b	22 ±1	31 ± 8	26 ± 3			
15	8.2±0.8	9.3±0.1	7.7±0.4			
16	24 ± 8	$10 \pm 0,5$	21 ± 5			
17	$7.9 \pm 3.4$	$8.1 \pm 0.7$	11 ± 1			
18	1.7 ±1.2	$0.57 \pm 0.16$	$0.86 \pm 0.7$			
19	33 ± 5	37 ±7	29 ±6			
20	224 ± 29	200 ± 30	240 ± 21			

<sup>&</sup>lt;sup>a</sup> Compound concentration required to inhibit cell proliferation by 50%.

Scheme 2. Synthesis of N-methoxy-indolo[2,3-a]carbazoles.

to the corresponding *N*-methoxy derivative **9** as a yellow solid (Rf 0.50 (CHCl<sub>3</sub>), m.p.  $234\sim236^{\circ}$ C (*n*-heptane-EtOAc). HR-MS calcd for  $C_{22}H_{15}N_3O_3$  M<sup>+</sup> 369.1113, found 369.1077). Under more drastic conditions (MeI, NaH, DMF) *N,N,O*-trimethyl derivative **10** was obtained (Scheme 2). Formation of **10** was also observed by TLC when the reaction done under milder conditions (MeI, acetone,  $K_2CO_3$ ) was allowed to stir overnight [**10**, a brown foam, Rf 0.22 (*n*-heptane - EtOAc, 10:1). HR-MS calcd for

 $\rm C_{23}H_{17}N_3O_3$  383.1270, found 383.1286].  $^1\rm H-$  and  $^{13}\rm C$  -NMR data for the compounds **9**, **10** are presenteed in the Tables 2 and 3 respectively.

 $N^{\text{ind}}$ -Alkoxy-arcyriarubin derivatives (N-alkoxy-bisindolylmaleimides  $13a\sim c$ ) were synthesized by the condensation of (indol-3-yl)acetamide with methyl 3-(1-alkoxyindol-3-yl)glyoxylates.<sup>8)</sup> (Scheme 3). 1-Alkoxyindoles  $11a\sim c$  were prepared from indoline by the oxidation with aqueous hydrogen peroxide in the presence of catalytic

Table 2. <sup>1</sup>H-chemical shifts ( $\delta$ , ppm) and coupling constants (J Hz) for compounds 9, 10, 13a $\sim$ c, and 14a, b (DMSO- $d_6$ ).

#	H2, H2'	H4, H4'	H5, H5'	H6, H6'	H7, H7'	N <sup>indole</sup> H	Other
9	-	9.4; 9.0;	7.46; 7.38;	7.66; 7.58;	7.82; 7.84;	12.42;	-ОС <u>Н</u> <sub>3</sub> 4.27, 3H, s;
		1H, d	1H, t	1H, t	1H, d;	1H, s	N-C <u>H</u> <sub>3</sub> 3.16, 3H, s;
10	•	9.14; 1H,	7.48; 1H, t,	7.66-7.72,	7.81, 1H,	-	-OC <u>H</u> <sub>3</sub> 4.28, 3H, s;
		d, J=8.02;	J=7.50	2H, complex	d, J=8.38;		N <sup>indole</sup> -C <u>H</u> <sub>3</sub> 3.15, 3H, s;
		9.08; 1H,	7.43, 1H, t,	,	7.78, 1H,		N <sup>mal</sup> -C <u>H</u> <sub>3</sub> 3.57; 3H, s
	,	d, J=7.96	J=7.49		d, J=8.05		
13a	7.9; 1H, s;	7.46; 1H,	7.12; 1H, t,	6.77, 1H, t,	6.96, 1H,	11.75;	OC <u>H</u> <sub>3</sub> 4.08, 3H, s;
	7.84; 1H,	d, J=7.86;	J=7.22	J=8.01;	d, J=8.14;	1H, s	N- <u>H</u> 11.0, 1H, s;
	d, J=2.74	7.41; 1H,	7.0, 1H, t,	6.65, 1H, t,	6.72, 1H,		
		d, J=8.06	J=7.5	J=7.1	d, J=7.55		
13b	7.83-7.58;	7.44, 1H,	7.12; 6.99;	6.73; 6.63;	6.97; 6.78;	11.75;	N <sup>mal</sup> - <u>H</u> , 10.96, 1H, s; -
	2H, m	d, J=7.80,	1H, t,	1H, t	1H, d	1H, s	OC <u>H</u> <sub>2</sub> - 4.2, 2H, t;
		7.39, 1H,					CH <sub>2</sub> C <u>H</u> <sub>2</sub> CH <sub>3</sub> 1.7, 2H,
		d, J=8,1					m;
							-CH <sub>2</sub> C <u>H</u> <sub>3</sub> 1.0, 3H, t.
13c	7.86, 1H, s	7.44, 1H,	7.10, 1H, t,	6.74, 1H, t,	6.94, 1H,	11.75,	-OC <u>H</u> <sub>2</sub> - 4.30, 2H, q;
	7.84, 1H,	d, J=8.24	J=7.63	J=7.52,	d, J=8.05	1H, d,	CH <sub>2</sub> C <u>H</u> <sub>3</sub> 1.28, 3H, t.
	d, J=2.78	7.38, 1H,	6.99, 1H, t,	6.62, 1H, t,	6.69, 1H,	J=2.56	N <sup>mal</sup> - <u>H</u> , 10.98, 1H, s;
		d, J=8.09	J=7.62	J=7.45	d, J=7.83		
14a	7.91; 7.81,	7.47; 1H,	7.14; 1H, t,	6.78; 1H, t,	6.97; 1H,	11.9;	OC <u>H</u> <sub>3</sub> 4.08, 3H, s;
	1H, s	d, J=7.5,	J=7.35;	J=8.0;	d; J=8.05;	1H, s;	N-C <u>H</u> <sub>3</sub> 3.06, 3H, s;
		7.40; 1H,	7.0; 1H, t,	6.68; 1H, t,	6.71, 1H,		
		d, J=8.2	J=7,3	J=8.1	d; J=8.15		
14b	7.95; 1H,	7.45; 7.4;	7.13; 6.95;	6.77; 6.63,	6.9; 6.69;	11.78;	N <sup>mal</sup> -C <u>H</u> <sub>3</sub> , 3.06, 3H, s;
	d, J=2.8;	1H, d	1H, t	1H, t	1H, d	1H, s;	-OC <u>H</u> <sub>2</sub> CH <sub>2</sub> - 4.2, 2H, t;
	7.85; 1H, s						-OCH₂C <u>H</u> ₂CH₃ 1.7,
							2H, m
							-OCH <sub>2</sub> CH <sub>2</sub> C <u>H</u> <sub>3</sub> 1.0,
							3H, t.

771

Table 3.  $^{13}$ C-chemical shifts for compounds 9, 10, and 12~14. (CDCl<sub>3</sub> for 10 and 12a~c, DMSO- $d_6$  for the rest)

Comp.	Alkyl	Carbonyl	Aromatic CH	Aromatic C	
9	N- <u>C</u> H <sub>3</sub> 23.6, O <u>C</u> H <sub>3</sub>	169.3, 169.5	108.8, 111.7, 112.2, 112	.7, 116.8, 118.3, 119.8, 120.3, 120.7,	
	65.7		121.9, 124.2, 124.7, 125.9, 127.0, 127.2, 127.4, 137.0, 141.4		
10	N- <u>C</u> H <sub>3</sub> 23.6,	169.3, 169.5	109.3, 110.7, 121.0,	118.1, 118.7, 118.8, 121.5, 121.6,	
	N <sup>ind</sup> - <u>C</u> H <sub>3</sub> 33.7,		122.9, 125.6, 125.9,	121.7, 127.4, 131.9, 140.5, 143.2,	
	N-O- <u>C</u> H <sub>3</sub> 65.7		127.6, 127.7,		
12a	O- <u>C</u> H <sub>3</sub> 52.7,	162.8, 176.5	108.8, 122.8, 124.0,	108.7, 123.3, 132.0	
	N-O- <u>C</u> H <sub>3</sub> 66.9		124.6, 133.1		
12b	- <u>C</u> H <sub>3</sub> 10.1, - <u>C</u> H <sub>2</sub> -	162.9, 176.5	108.8, 122.7, 123.9,	108.5, 123.4, 132.6	
	21.5,		124.5, 133.8		
	O- <u>C</u> H <sub>2</sub> - 81.2, O- <u>C</u> H <sub>3</sub>				
	52.6				
12c	CH <sub>2</sub> CH <sub>3</sub> 13.6,	162.8, 176.5	108.8, 122.6, 123.9,	108.5, 123.3, 132.7	
	O- <u>C</u> H <sub>2</sub> - 75.3,		124.5, 133.9	·	
	O- <u>C</u> H <sub>3</sub> 52.6				
13a	O- <u>C</u> H <sub>3</sub> 52.7,	172.6, 172.7	108.3, 111.8, 119.5,	101.8, 105.2, 122.0, 124.9, 125.9,	
	N-O- <u>C</u> H <sub>3</sub> 66.9		120.2, 120.9, 121.5,	129.4, 131.1, 136.1	
			121.7, 122.7, 126.1,		
			129.7		
13b	- <u>C</u> H <sub>3</sub> 10.0, - <u>C</u> H <sub>2</sub> -	172.5, 172.7	108.3, 111.8, 119.3,	101.7, 105.2, 121.9, 124.9, 125.9,	
	20.9,		120.1, 120.7, 121.4,	129.4, 131.7, 136.0	
	O- <u>C</u> H <sub>2</sub> - 80.0		121.6, 122.5, 126.5,		
			129.6		
13c	CH <sub>2</sub> CH <sub>3</sub> 13.4,	172.7, 172.5	108.4, 111.8, 119.3,	101.6, 105.1, 121.8, 124.9, 125.9,	
	O- <u>C</u> H <sub>2</sub> - 74.2,		120.0, 120.7, 121.3,	129.3, 131.8, 136.0	
			121.6, 122.5, 126.7,		
			129.6		
14a	N-O- <u>C</u> H <sub>3</sub> 66.5,	171.4, 171.5	108.3, 111.9, 119.5,	101.8, 105.2, 121.9, 124.8, 125.2,	
	N- <u>C</u> H <sub>3</sub> 24.0		120.2, 120.9, 121.5,	128.9, 131.1, 136.1	
			121.7, 122.7, 126.1,		
			129.7		
14b	- <u>C</u> H <sub>3</sub> 10.0, - <u>C</u> H <sub>2</sub> -	171.4, 174.5	108.5, 111.9, 119.4,	101.7, 105.2, 121.9, 124.8, 125.3,	
	21.0,		120.2, 120.8, 121.4,	128.9, 131.7, 136.1	
	O- <u>C</u> H <sub>2</sub> - 80.1, N- <u>C</u> H <sub>3</sub>		121.7, 122.7, 126.6,		
	24.0		129.7		
			·		

Scheme 3. Synthesis of *N*-alkoxyderivatives of arcyriarubin A.

Fig. 2.

amounts of  $Na_2WO_4\times 2H_2O$  under PTC conditions (triethylbenzylammonium chloride was used as a phase transfer agent) with subsequent alkylation. The modification of this method of indoline *N*-oxidation proved to be more convenient than the oxidation in aqueous methanol followed by alkylation, as it was first described by SOMEI,  $^{7)}$  since oxidation and alkylation were performed as a one-pot reaction. 1-Methoxyindole (11a), 1-propoxyindole (11b), and 1-ethoxyindole (11c) were obtained as colorless oils in 40% yields.

The corresponding methyl 3-(1-alkoxyindol-3-yl)-glyoxylates 12a~c were prepared by the reaction of 11a~c with oxalyl chloride and subsequent treatment with methanolic triethylamine solution. The mixture of a methyl 3-(1-alkoxyindolyl)-glyoxylate (12a, b or c) with indolyl-3-acetamide in THF was then treated with KOBu' solution in

THF to achieve the final cyclization to give after column chromatography the corresponding bis-indolylmaleimides in 23~25% yields: 3-(indol-3-yl)-4-(1-methoxy-indol-3yl)-pyrrole-2,5-dione (13a) [Rf 0.24 (n-heptane - EtOAc, 3:2), HR-MS calc. for  $C_{21}H_{15}N_3O_3$  357.1113 found 357.1091], 3-(indol-3-yl)-4-(1-propoxy-indol-3-yl)pyrrole-2,5-dione (13b) [Rf 0.29 (*n*-heptane - EtOAc, 3:2), HR-MS calc. for  $C_{23}H_{19}N_3O_3$  385.1426 found 385.1401] or 3-(indol-3-yl)-4-(1-ethoxy-indol-3-yl)-pyrrole-2,5-dione (13c) [Rf 0.27 (n-heptane - EtOAc, 3:2), HR-MS calc. for  $C_{10}H_{17}N_3O_3$  371.1270, found 371.1259]. Bis-indolylmaleimides 13a, b were then converted into  $N^{\text{mal}}$ -methyl derivatives 14a, b in 90% yields by the treatment with MeI and K<sub>2</sub>CO<sub>3</sub> in acetone (Scheme 3): [ 3-(Indol-3-yl)-4-(1methoxy-indol-3-yl)-1-methyl-pyrrole-2,5-dione (14a), Rf 0.36 (*n*-heptane - EtOAc, 3:2), HR-MS calc.

 $\rm C_{22}H_{17}N_3O_3$  371.1269 found 371.1272], and 3-(indol-3-yl)-4-(1-propoxy-indol-3-yl)-1-methyl-pyrrole-2,5-dione (**14b**) [Rf 0.45 (*n*-heptane - EtOAc, 3:2), HR-MS calcd for  $\rm C_{24}H_{21}N_3O_3$  399,1582 found 399,1575].  $^1\rm H-$  and  $^{13}\rm C-NMR$  data for the compounds **13a**, **b**, **c** and **14a**, **b** are presenteed in the Tables 2 and 3 respectively.

To compare the role of an alkoxy and alkyl substituent at the indole nitrogen atom for the cytotoxic activity, 6-methyl-indolo[2,3-a]pyrrolo[3,4-c]carbazol-5,7-dione 15, and bisindolylmaleimides  $16\sim20$  were synthesized as previously described (Fig. 2).<sup>4,8)</sup>

Cytotoxicity of the compounds obtained was evaluated in three cell cultures (Table 1). The comparison of compounds 9, 10 and 15 shows that the introduction of  $N^{\text{ind}}$ -MeO group into 15 leads to some decrease of cytotoxicity and further N-alkylation of the second indole nucleus (10) results in a dramatic decrease of cytotoxic properties. This can be explained by the role of hydrogen bonds in the biological activity of these compounds. Comparison of the cytotoxic activities of bis-indolylmaleimides (13~20) demonstrates that the unsymmetrical  $N^{\text{ind}}$ -alkoxy derivatives (13a $\sim$ c) and  $N^{\text{ind}}$ -alkyl derivative (18) are more cytotoxic than the unsubstituted bisindolylmaleimide 16, the most cytotoxic in this series are  $N^{\text{ind}}$ -ethoxy (13c) and  $N^{\text{ind}}$ -ethyl (18) compounds. The substitution of the nitrogen in the maleimide ring (14a, b, 19) leads to a decrease of cytotoxicity; and the IC<sub>50</sub> values for compounds 14a, b, and 19 are of the same order. The substitution of the second indole nitrogen (20) again leads to dramatic decrease in activity. However comparison of compounds 16 and 17 shows that mono N-methylation of the maleimide ring does not influence negatively cytotoxic properties perhaps because 17 still has two N-H bonds. It is interesting to note that the maleimide derivative containing three indole nuclei (5) was found to be rather cytotoxic. All the compounds investigated demonstrated no antiviral activity [against HIV-1, HIV-2, HCMV (compounds 2, 10, 13a and 15) and varicella-zoster virus (compounds 9, 10, 13a and 15) and herpes simplex virus type 1 (KOS), and type 2 (G), vaccinia virus, vesicular stomatitis virus, Coxsackie, Sindbis, Punta Toro virus, reovirus-1, parainfluenza-3 and respiratory syncytial virus 13b, 14a, b, 16~20] at subtoxic

concentrations, that is at concentrations barely lower than the overtly cytotoxic concentrations.

## Acknowledgements

This work was supported by the Russian Fund for Fundamental Research, grant number 01-03-33028 and the "Fonds voor Wetenschappelijk Onderzoek-Vlaanderen", grant number G.0104.98.

## References

- Bush, J. A.; B. H. Long, J. J. Catino & W. T. Bradner: Production and biological activity of rebeccamycin, a novel antitumor agent. J. Antibiotics 40: 668~678, 1987
- PINDUR, U.; Y.-S. KIM & F. MEHARBANI: Advances in indolo[2,3-a]carbazole chemistry: design and synthesis of protein kinase C and topoisomerase I inhibitors. Curr. Med. Chem. 6: 29~68, 1999
- 3) MELNIK, S. Ya.; A. A. BAKHMEDOVA, L. D. GARAEVA, O. V. GORYUNOVA, T. D. MINIKER, I. L. PLIKHTYAK, T. P. IVANOVA & I. V. YARTSEVA: Synthesis and antiproliferative properties of bis(indolyl)-1H-furan-2,5-dione and bis(indolyl)-1H-pyrrole-2,5-dione N-glycosides. Bioorganicheskaya Khimiya (Russ.) 22: 458~461, 1991
- 4) SLATER, M. J.; R. BAXTER, R. W. BONSER, S. COCKERILL, K. GOHIL, N. PARRY, E. ROBINSON, R. RANDALL, C. YEATES, W. SNOWDEN & A. WALTERS: Synthesis of Nalkyl substituted indolocarbazoles as potent inhibitors of human cytomegalovirus replication. Bioorg. Med. Chem. Lett. 11: 1993~1995, 2001
- 5) Brenner, M.; H. Rexhausen, B. Steffan & W. Steglich: Synthesis of arcyriarubin A and related bisindolylmaleimides. Tetrahedron 44: 2887~2892, 1988
- 6) CHISHOLM, J. D.; J. GOLIK, B. KRISHNAN, J. A. MATSON & D. L. VAN VRANKEN: A caveat in the application of the excition chirality method to N,N-dialkyl amides. Synthesis and structural revision of AT2433-B1. J. Am. Chem. Soc. 121: 3801~3802, 1999
- SOMEI, M.: 1-Hydroxyindoles. Heterocycles 50: 1157~ 1210, 1999
- 8) FAUL, M. M.; L. L. WINNEROSKI & C. A. KRUMRICH: A new efficient method for the synthesis of bisin-dolylmaleimides. J. Org. Chem. 63: 6053~6058, 1998
- 10) CIAMICIAN, G. L. & P. SILBER: Ueber einige Derivate des Succinimides. Chem. Ber. 17: 553~557, 1884
- YUDINA, L. N.: Synthesis and study of the properties of indolocarbazoles and ascorbigenes. Ph. D. Thesis, Gause Institute of New Antibiotics of the Russian Academy of Medical Sciences, Moscow, 2000